## Mechanistic Study of BPh<sub>3</sub>-Catalyzed *N*-Methylation of Amines with CO<sub>2</sub> and Phenylsilane

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## Abstract

BPh<sub>3</sub> was found to catalyze the highly selective N-methylation of secondary amines at 30 <sup>o</sup>C with CO<sub>2</sub> (1 atm) and PhSiH<sub>3</sub> under solvent-free conditions [1]. In the present work, we have clarified the mechanism of BPh<sub>3</sub>-catalyzed N-methylation of N-methylaniline with CO<sub>2</sub> and PhSiH<sub>3</sub> using the DFT calculations at  $\omega$ B97XD/6-31G(d), 6-31++G(d,p) for the hydride of PhSiH<sub>3</sub> level of theory [2]. The DFT results revealed that the BPh<sub>3</sub> promotes the conversion of N-methylaniline and CO<sub>2</sub> substrates into a zwitterionic carbamate to give three-component species  $[Ph(Me)(H)N^+CO_2^-\cdots BPh_3]$ . The carbamate and BPh<sub>3</sub> act as the nucleophile and Lewis acid, respectively, for the cooperative activation of PhSiH<sub>3</sub> to generate borohydride [HBPh<sub>3</sub>]<sup>-</sup> species, which catalyzes the reduction of CO<sub>2</sub> to form key reactive species such as silvl formats, bis(silvl)acetals, and formaldehyde. These key CO<sub>2</sub>-derived reactive species are essential for N-methylation reaction. In addition, we realized that a water molecule might act as a nucleophile to activate PhSiH<sub>3</sub>. Hence, we have explored other mechanisms and suggested hypothetical water-assisted mechanisms for the generation of active [HBPh<sub>3</sub>]<sup>-</sup> species. Our results indicate that [HBPh<sub>3</sub>]<sup>-</sup> species can be generated relatively easily by the involvement of water. Interestingly, we have found that if acetonitrile is used as a solvent, acetonitrile can serve as a Lewis base to weakly activate carbonic acid or water, both of which can attack the silicon atom of PhSiH<sub>3</sub>. The elucidation of this catalytic mechanism will be useful for the further progress of CO<sub>2</sub> fixation chemistry in future and development of new catalysts and reactions.

## References

[1] Murata, T.; Hiyoshi, M.; Maekawa, S.; Saiki, Y.; **Ratanasak, M.;** Hasegawa, J.; Ema, T.\*, Deoxygenative  $CO_2$  Conversions with Triphenylborane and Phenylsilane in the Presence of Secondary Amines or Nitrogen-Containing Aromatics. *Green Chem.* **2022**, *24* (6), 2385-2390.

[2] Ratanasak, M.; Murata, T.; Adachi, T.; Hasegawa, J.\*; Ema. T.\*, Mechanism of BPh<sub>3</sub>-Catalyzed *N*-Methylation of Amines with CO<sub>2</sub> and Phenylsilane: Cooperative Activation of Hydrosilane. *Chem. Eur. J.* 2022, e20220210 (Selected as Inside Front Cover & Selected as a Hot Paper).